

SELF STUDY MODULE

Liquefaction in everyday practice and its impacts

Objective

Liquefaction processes and their applications

Vocabulary

Simple Liquefaction Process: A single pass process to which a gas enters and a vapor and a liquid stream leaves.

Linde Process: A process to which a gas enters, and only a liquid leaves, the vapor is recycled and is mixed with the feed stream.

Throttling: Rapid decrease of pressure of a real fluid to cause sharp temperature decrease or liquefaction.

Useful diagrams

PH diagram of nitrogen

PH diagram of methane

PH diagram of CO₂

Balance equations

The general mass conservation law $\frac{dm}{dt} = \sum_{in} m_i - \sum_{out} m_j$

The general conservation of energy or the first law of thermodynamics

$$\frac{dU}{dt} = \sum_{in} m_i h_i - \sum_{out} m_j h_j + \dot{Q} + \dot{W}_S - P \frac{dv}{dt}$$

The entropy balance has a generation term

$$\frac{dS}{dt} = \sum_{in} m_i s_i - \sum_{out} m_j s_j + \frac{\dot{Q}}{T} + \dot{S}_{gen}$$

Calculate

1. A group of COVID 19 vaccines require -80 C as storage medium. You are asked to design the system for this process. Dry ice seems a feasible domain. Find the PH diagram of CO2 and propose the flow chart for this system. How much energy is needed? How much cooling is required?
2. Design a liquefaction process for natural gas transportation. How much work is needed/kg and how much cooling is required/kg natural gas that is liquefied? Why do we liquefy natural gas?

Bibliography

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