

## ***SELF STUDY MODULE***

### ***Liquefaction in everyday practice and its impacts***

#### ***Objective***

Liquefaction processes and their applications

#### ***Vocabulary***

**Simple Liquefaction Process:** A single pass process to which a gas enters and a vapor and a liquid stream leaves.

**Linde Process:** A process to which a gas enters, and only a liquid leaves, the vapor is recycled and is mixed with the feed stream.

**Throttling:** Rapid decrease of pressure of a real fluid to cause sharp temperature decrease or liquefaction.

#### ***Useful diagrams***

PH diagram of nitrogen

PH diagram of methane

PH diagram of CO<sub>2</sub>

#### ***Balance equations***

The general mass conservation law  $\frac{dm}{dt} = \sum_{in} m_i - \sum_{out} m_j$

The general conservation of energy or the first law of thermodynamics

$$\frac{dU}{dt} = \sum_{in} m_i h_i - \sum_{out} m_j h_j + \dot{Q} + \dot{W}_S - P \frac{dv}{dt}$$

The entropy balance has a generation term

$$\frac{dS}{dt} = \sum_{in} m_i s_i - \sum_{out} m_j s_j + \frac{\dot{Q}}{T} + \dot{S}_{gen}$$

### **Calculate**

1. A group of COVID 19 vaccines require -80 C as storage medium. You are asked to design the system for this process. Dry ice seems a feasible domain. Find the PH diagram of CO2 and propose the flow chart for this system. How much energy is needed? How much cooling is required?
2. Design a liquefaction process for natural gas transportation. How much work is needed/kg and how much cooling is required/kg natural gas that is liquefied? Why do we liquefy natural gas?

### **Bibliography**

- S. Sandler Chemical Biochemical and Engineering thermodynamics, 4<sup>th</sup> edition, Wiley
- M. Koretsky, Engineering and Chemical Thermodynamics, 2<sup>nd</sup> edition, Wiley, 2013, NY.
- M.J. Moran, H. N. Shapiro, D.D. Boettner, M.B. Bailey, Principles of Engineering Thermodynamics, 7<sup>th</sup> edition, John Wiley and Sons, 2012, NY.