

### PHYS-430 RECITATIN 3

1.) A quantum harmonic oscillator with frequency  $\omega$  can exist in states with energies (relative to zero point energy) of  $\hbar\omega$ ,  $2\hbar\omega$ ,  $3\hbar\omega$ , ... . If there are  $N$  identical oscillators, the number of ways to obtain an energy  $n\hbar\omega$  is

$$g(N, n) = \frac{(N+n-1)!}{n!(N-1)!}$$

- (a) Find the entropy of this system of oscillators when  $N$  is large (so that you can use the Stirling approximation and you can ignore 1 compared to  $N$ ).
- (b) Find an expression for the energy in terms of the temperature.

2.) Consider a system composed of  $N$  non-interacting, distinguishable, spin-1/2 particles. In an applied external magnetic field  $B$  the “up” and “down” states of each spin have respectively energies  $-\mu B$  and  $+\mu B$  where the magnetic moment  $\mu$  is a constant.

- (a) What is the total number of states of this system?
- (b) How many states are available to the system if there is an applied  $B$  and if the system's total energy is  $E$  where  $-N\mu B < E < N\mu B$  and  $E$  is an integer multiple of  $\mu B$ .
- (c) Write the entropy of the system as a function of  $E$ ,  $B$ , and  $N$ . Make use of Stirling's formula to express any factorials that appear in your expressions.
- (d) Calculate the temperature  $T$  of the system as a function of  $E$ ,  $B$ , and  $N$ .
- (e) Solve for the energy  $E$  as a function of the temperature  $T$ ,  $B$ , and  $N$ .

3.) Consider a lattice with  $N$  spin-1 atoms. Each atom can be in one of three spin states,  $S_z = -1; 0; +1$ . Let  $n_{-1}$ ,  $n_0$ , and  $n_1$  denote the respective number of atoms in each of those spin states. Assume that no magnetic field is present, so all atoms have the same energy.

- (a) Find the total entropy as a function of  $n_{-1}$ ,  $n_0$ , and  $n_1$
- (b) Which configuration ( $n_{-1}; n_0; n_1$ ) maximizes this entropy?
- (c) What is the entropy in this maximized configuration?